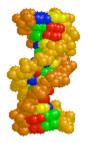
VISUALIZING MACROMOLECULES - JMOL COMPUTER ACTIVITY

Go to the website: http://www.biotopics.co.uk/JmolApplet/jcontentstable.html



Manipulations of molecules:

- ✓ To rotate the molecule yourself, **left-click and drag** on it. Drag the cursor up and down for x-axis rotation, left-right for y-axis rotation.
- ✓ To translate the molecule yourself, **shift-double-click and drag** on the structure--the molecule will follow the mouse.
- \checkmark To zoom the molecule yourself, **shift-click and drag**. Drag the cursor down to zoom in, up to zoom out.
- ✓ TO RESET THE MOLECULE, SHIFT-DOUBLE-CLICK MOUSE

Make sure to view the following molecules and answer the questions below:

Let's start with carbohydrates...

1. Click GLUCOSE. Practice <u>zooming in</u> and <u>rotating</u> the glucose molecule. Click the "label glucose" button and <u>label the carbon atoms</u> (C1, C2, etc.) on the diagram.

2. Click on ALPHA-GLUCOSE VS. BETA-GLUCOSE

Practice <u>zooming in</u> and <u>rotating</u> the alpha glucose molecule.

How does α (alpha) and β (beta) glucose differ structurally?

3. Click on GLUCOSE-GALACTOSE COMPARISON

Galactose and glucose are considered to be ______, which means that they both share the same formula $C_6H_{12}O_6$.

What is the structural difference between them?

4. Click on SUCROSE

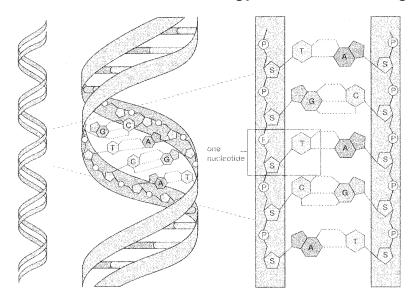
Sucrose is a disaccharide - formula ______ - consisting of one _____ unit combined with one _____ unit.

5. Click on AMYLOSE Amylose is the name given to linear sections of the starch molecule.
What are the linkages called between the glucose units? (be specific)
This model shows 36 glucose units, forming a helical structure - effectively a tube. Click the appropriate button to move to the end to see the structure.
6. Click on CELLULOSE
This model shows 9 glucose units joined by alternating bonds.
Notice the shape of the chain. Describe the difference in shape of the cellulose chain with that of amylose.
Let's study lipids now
7. Click on GLYCEROL
What is the functional group in this molecule?
8. Click on SATURATED AND UNSATURATED FATTY ACIDS
Locate the carboxyl group of stearic acid. What colour represents the oxygen atoms?
What is the chemical formula of stearic acid?
What is the chemical formula of oleic acid?
Why is there a difference in the number of hydrogen atoms?
9. Click on TRIGLYCERIDES (FATS) Zoom into the molecule.
What is the bond called between the fatty acids and the central glycerol unit?

Let's turn to proteins now...

On to levels of protein structure (General Principles)
10. Click on PRIMARY STRUCTURE There are 20 different amino acids.
For a dipeptide (i.e. 2 amino acids linked together), how many different amino acid combinations are possible?
For a tripeptide, how many different combinations are possible?
Determine the number of different combinations of amino acid sequences that can be possible for a peptide with "n" amino acids
Clearly the number of possible combinations is almost infinite when larger numbers of amino acids are combined to form a polypeptide.
Secondary structure
The polypeptide chain can fold back on itself in a number of ways. Click the "BACKBONE" button to show the backbone for both secondary structures.
One way is to fold into a helical structure, and the
other is to fold as asheet.
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Finally, a look at nucleic acids... Recall from Gr. 11 Biology, the structure of a general nucleotide.



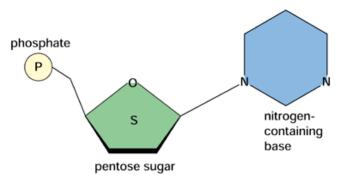


Figure 1.20. Generalized nucleotide. Nucleotides consist of a five-carbon simple sugar (ribose in the case of RNA and deoxyribose in DNA), a nitrogen base, and a phosphate group.

12. Click on DNA		
The outside edges of the DNA double helix are composed of alternating		
and groups. Click the appropriate button to see the sugar-phosphate backs	oone.	
What element does the orange atom represent?		
Across the middle of the molecule are the nitrogenous bases, shatoms in blue. In each case the bases do not directly contact the opposite strand, to which they are held by	heir partner on the	
Click the appropriate button to see the hydrogen bonds .		
13. Click on DNA Bases		
Which nitrogenous bases are purines?		
Which nitrogenous bases are pyrimidines?		
What is the structural difference between purines and pyrimidin		
How many hydrogen bonds are between A-T?		
You have now completed your task. Make sure you keep this sh	eet for future reference.	