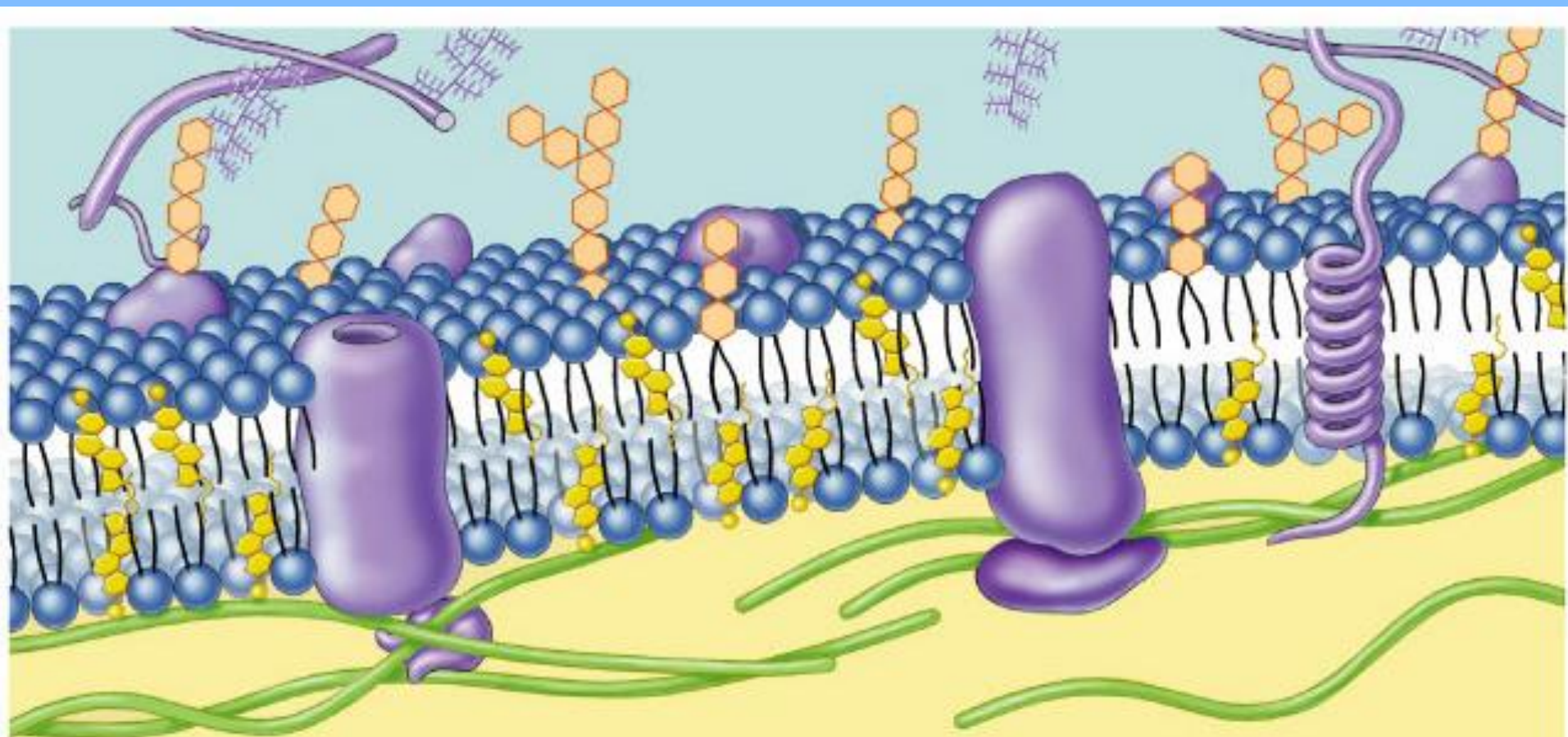


Membrane Transport

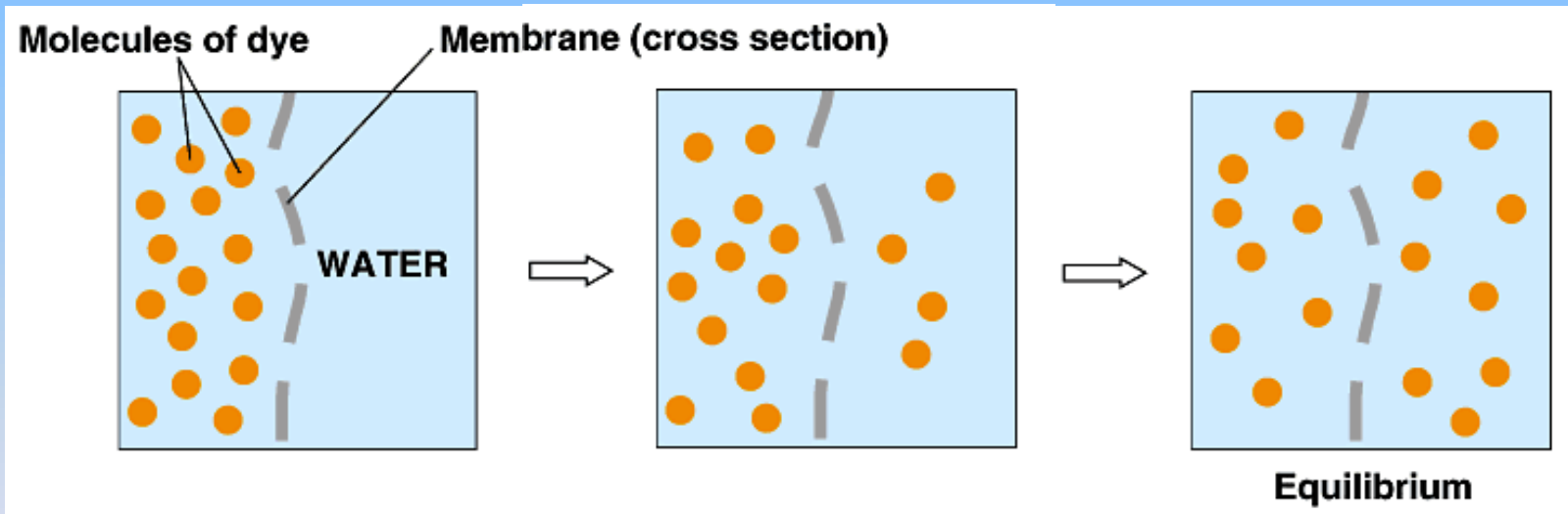


Passive Transport

- The movement of materials across a cell membrane without the use of energy (ATP)
- Movement of particles along a concentration gradient → from an area of higher concentration to an area of lower concentration (WITH a concentration gradient)

Diffusion

- 2nd Law of Thermodynamics governs biological systems
 - universe tends towards disorder (entropy)

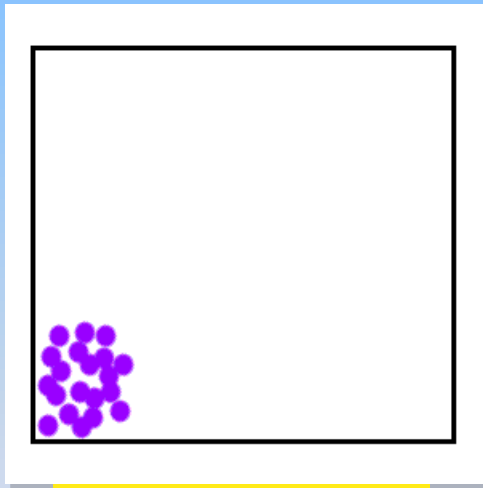


■ Diffusion

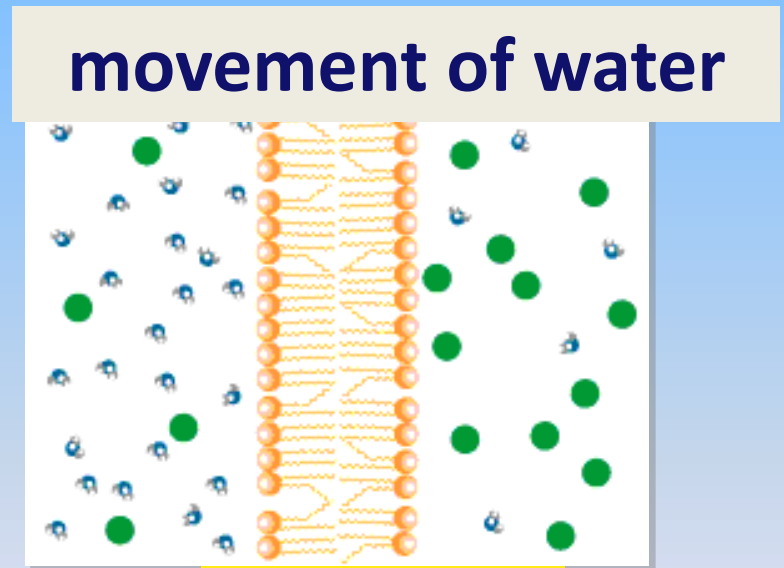
- ◆ movement from **HIGH** → **LOW** concentration

Simple Diffusion

- Movement of particles from **HIGH** to **LOW** concentration areas
 - “passive transport”
 - no energy needed
 - Will continue until equilibrium is achieved



diffusion

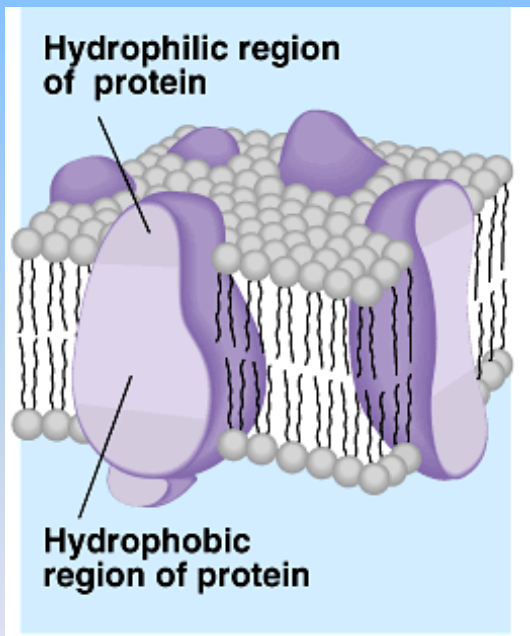


osmosis

Example: Movement of O_2 into cells and CO_2 out of cells

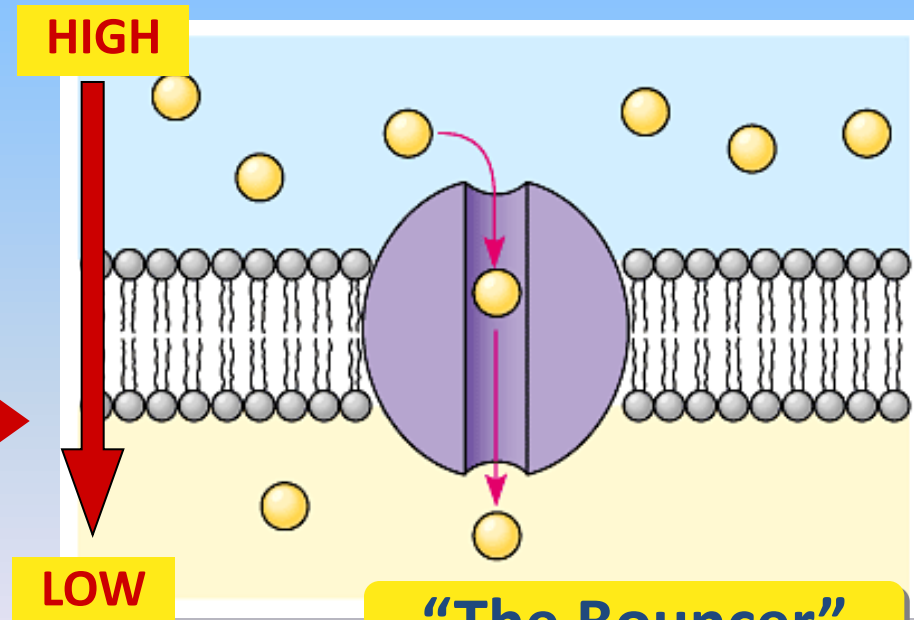
Facilitated Diffusion

- Molecules too large for simple diffusion or are hydrophilic
- Diffusion through protein channels
 - channels move specific molecules across cell membrane
 - no energy needed



facilitated = with help

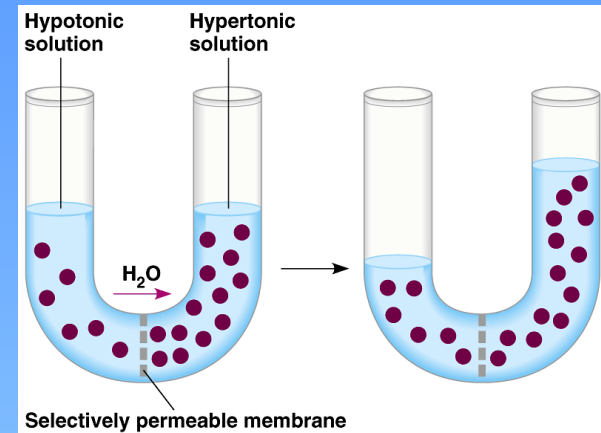
open channel = fast transport



Example: Glucose is constantly being used up inside the cell so the concentration gradient is maintained; Small ions

Osmosis

- Water is very important to life, so we talk about water separately
- Diffusion of **WATER** through a selectively permeable membrane from an area of *HIGH concentration* of water to an area of *LOW concentration* of water
 - Water moves from an area of lower solute concentration to an area of higher solute concentration

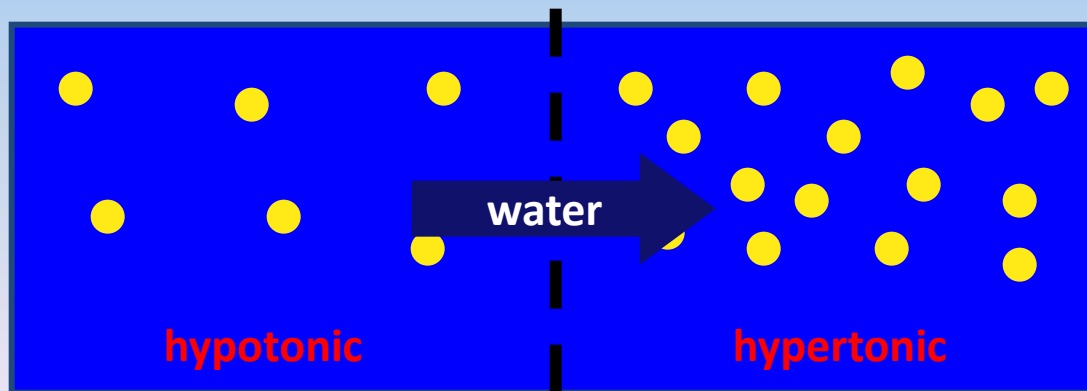


Osmosis is important to aquatic organisms.

Kidney's use osmosis to regulate water balance in the blood.

Concentration of water

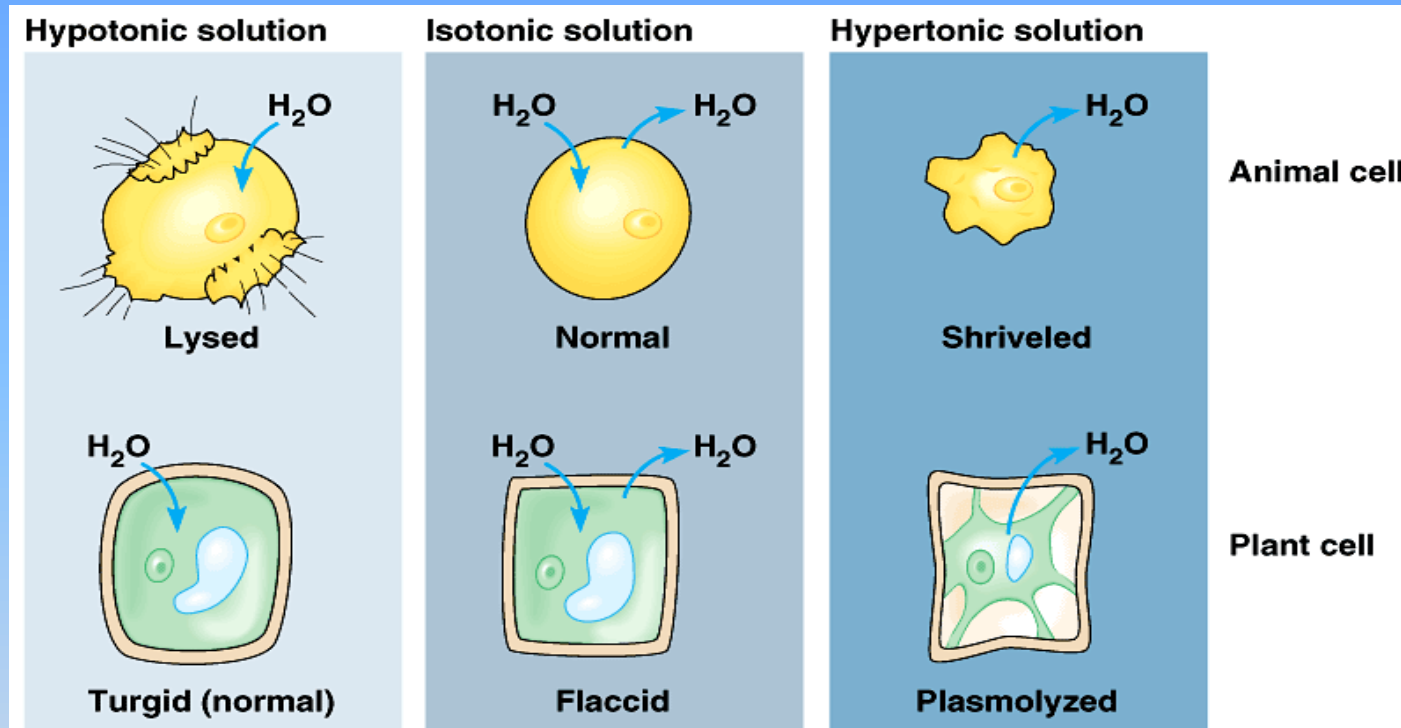
- Direction of osmosis is determined by comparing total solute concentrations
 - Hypertonic - more solute, less water
 - Hypotonic - less solute, more water
 - Isotonic - equal solute, equal water



net movement of water

Managing water balance

- Cell survival depends on balancing water uptake & loss



freshwater

In a hypotonic solution the concentration of solutes is higher inside the cell. Water moves into the cell

balanced

In a isotonic solution the concentration of solutes is equal inside the cell and out.

saltwater

In a hypertonic solution the concentration of solutes is higher outside the cell. Water moves out of the cell

1

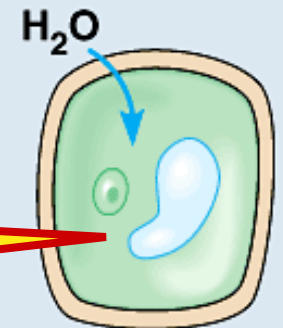
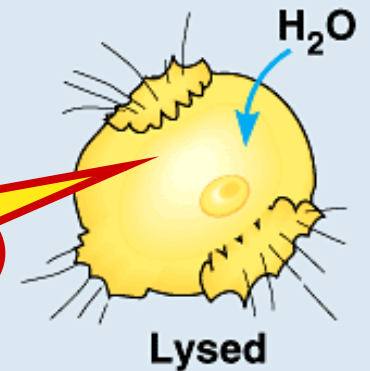
Managing water balance

- Hypotonic
 - a cell in fresh water
 - high concentration of water around cell
 - problem: cell gains water, swells & can burst
 - example: Paramecium
 - solution: contractile vacuole
 - pumps water out of cell
 - Uses ATP
 - plant cells
 - turgid = full
 - cell wall protects from bursting

KABOOM!

No problem, here

Hypotonic solution



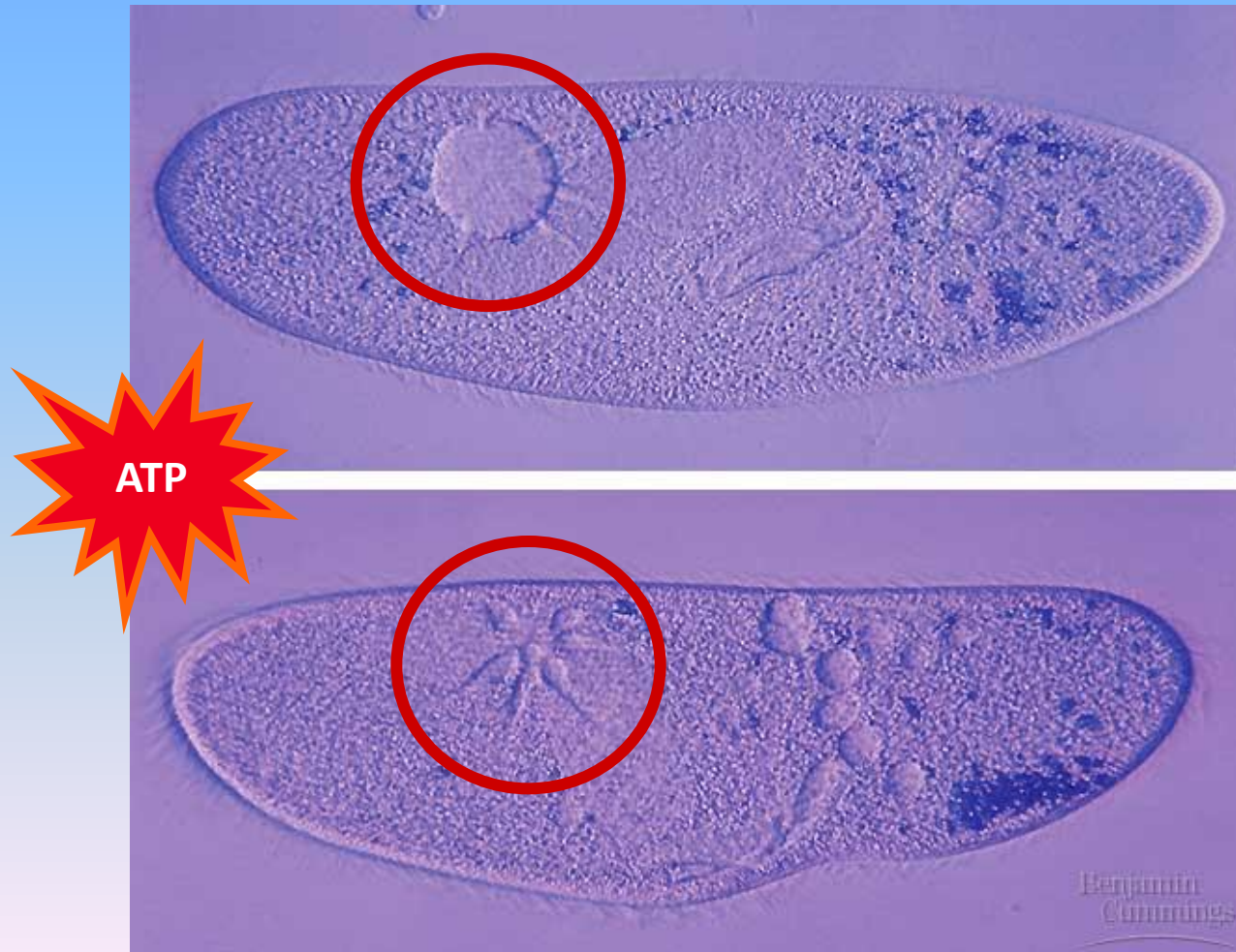
Turgid (normal)

freshwater

ATP

Pumping water out

- Contractile vacuole in *Paramecium*



2

Managing water balance

- Hypertonic

- a cell in salt water

- low concentration of water around cell

- problem: cell loses water & can die

- example: shellfish

- solution: take up water or pump out salt

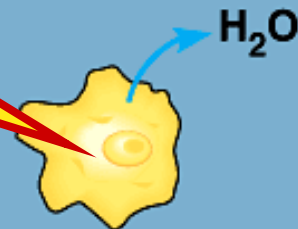
- plant cells

- plasmolysis = wilt

- can recover

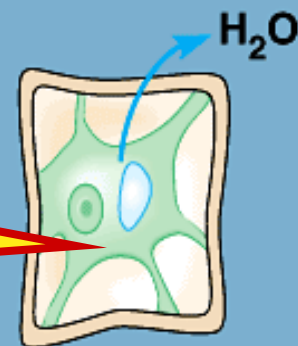
I'm shrinking,
I'm shrinking!

Hypertonic solution



Shriveled

I will
survive!



Plasmolyzed

saltwater

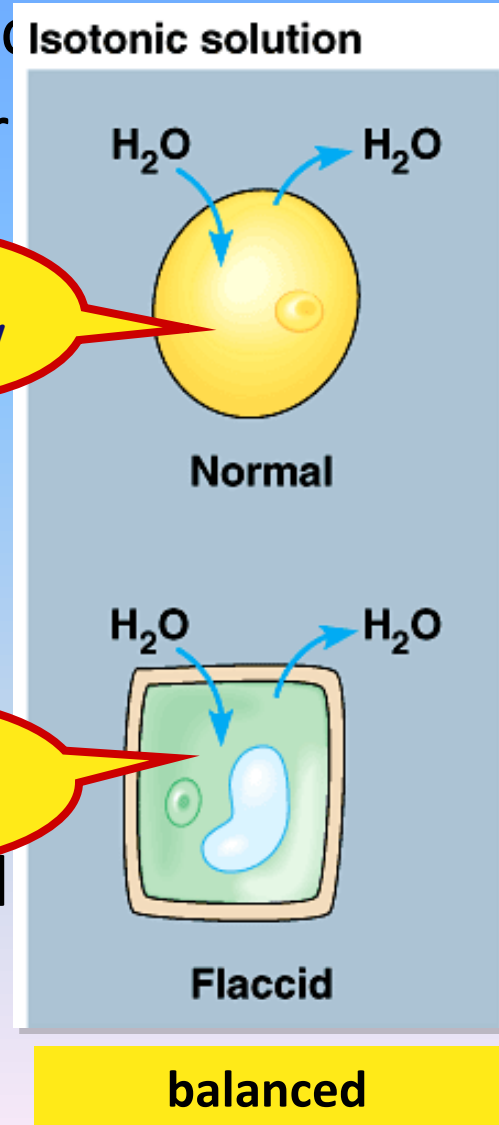
3

Managing water balance

- Isotonic
 - animal cell immersed in mild salt solution
 - no difference in concentration of water between cell & environment
 - problem: none
 - no net movement of water
 - cell in equilibrium
 - volume of cell is stable
 - example:
 - blood cells in blood plasma
 - slightly salty IV solution in hospital

That's perfect!

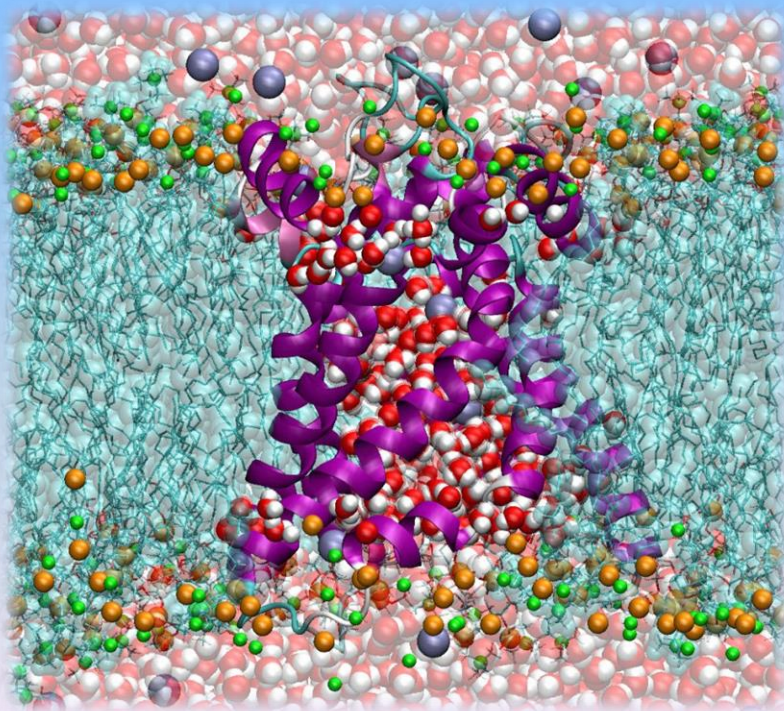
I could be better...



Aquaporins

1991 | 2003

- Water moves rapidly into & out of cells
 - evidence that there were water channels
 - protein channels allowing flow of water across cell membrane

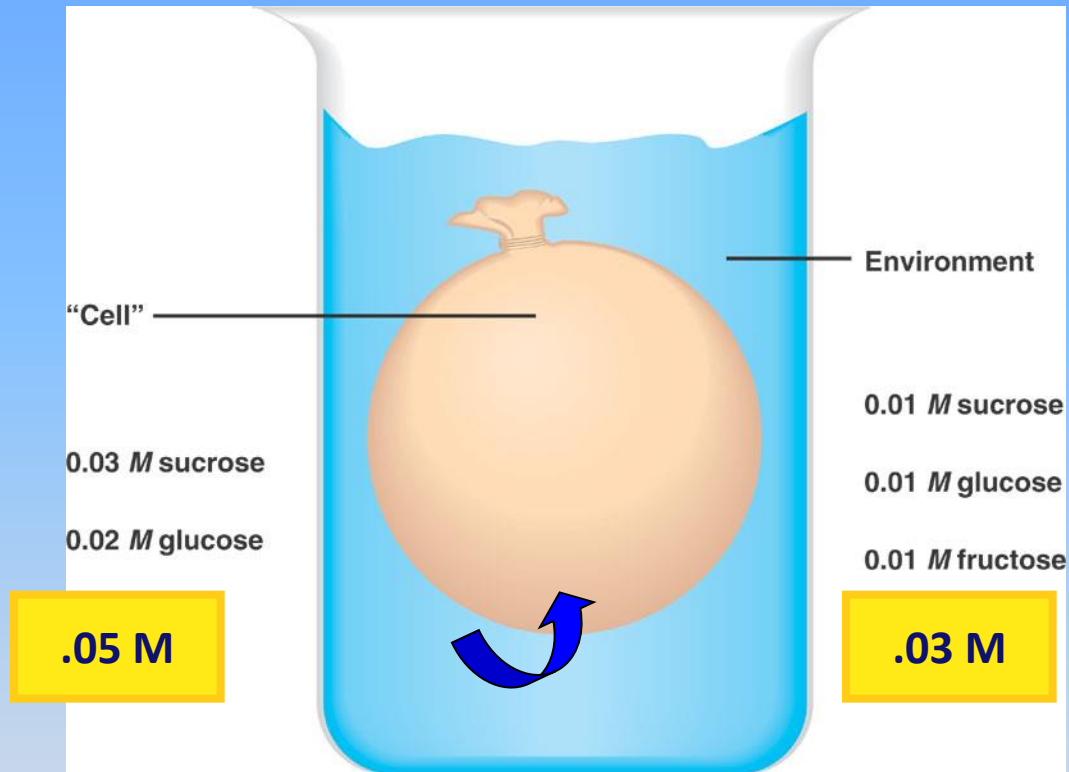


Peter Agre
John Hopkins



Roderick MacKinnon
Rockefeller

Do you understand Osmosis...



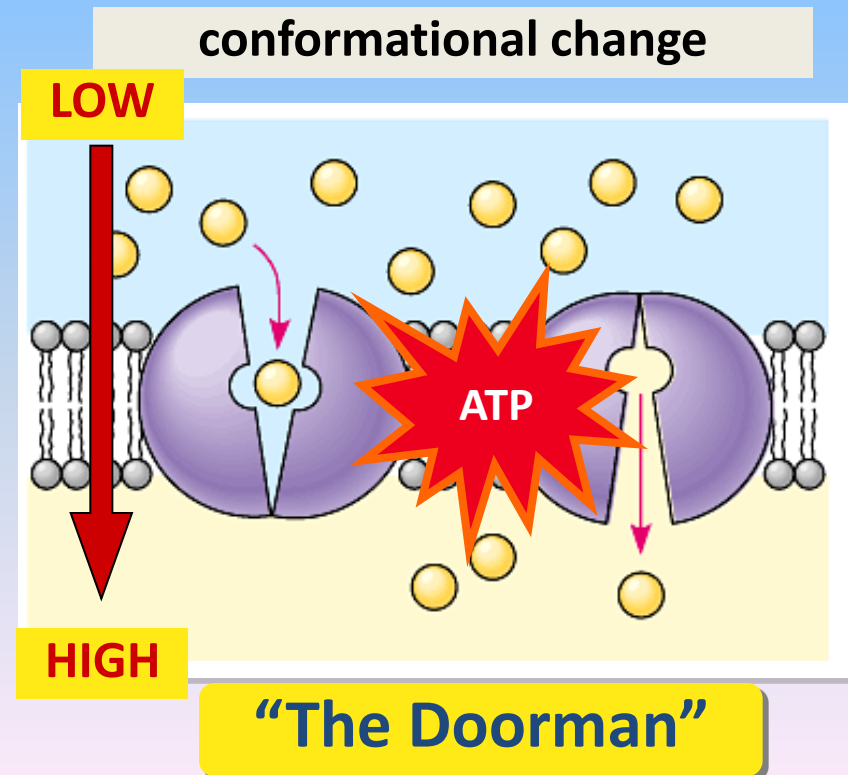
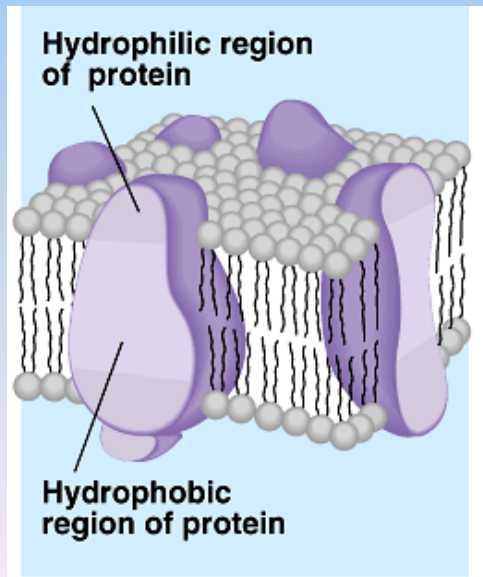
Cell (compared to beaker) → hypertonic or hypotonic

Beaker (compared to cell) → hypertonic or hypotonic

Which way does the water flow? → in or out of cell

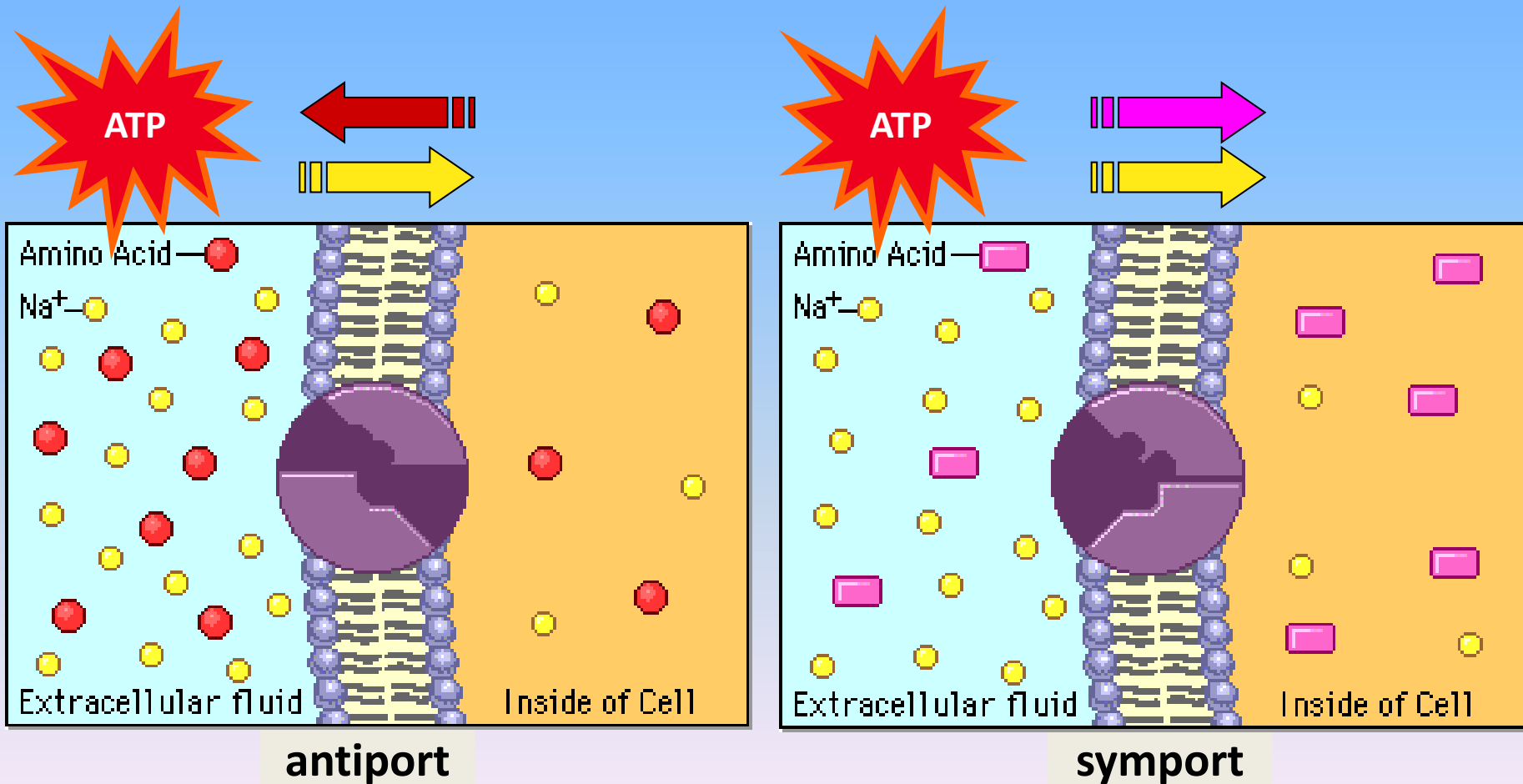
Active Transport

- Uses ATP (energy) to move materials across a cell membrane against a concentration gradient.
 - conformational shape change transports solute from one side of membrane to other
 - protein “pump”
 - “costs” energy = ATP



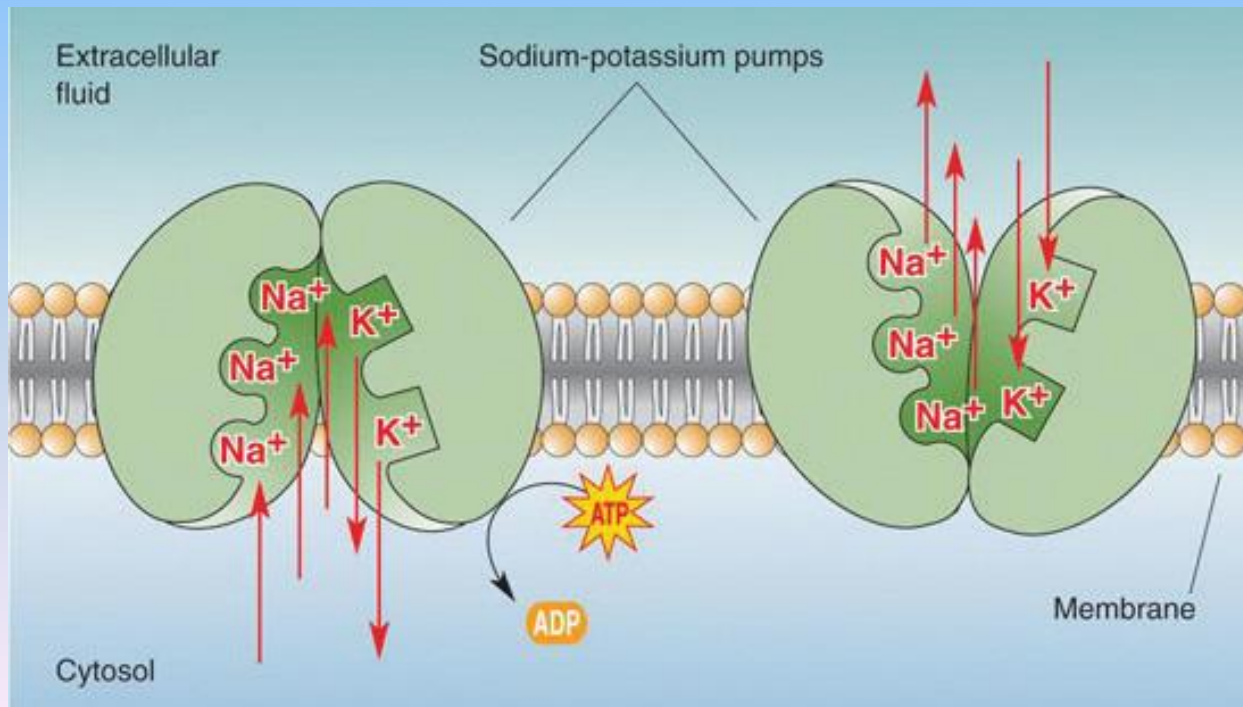
Active transport

- Many models & mechanisms



Na⁺/K⁺ pump

- Membrane transport proteins pump Na⁺ out of the cell and K⁺ into the cell against the concentration gradient. Uses ATP!
- Nerve cells must maintain a higher concentration of Na⁺ outside the cell and K⁺ inside the cell to function.



Getting through cell membrane

- Passive Transport

- Simple diffusion

- nonpolar, hydrophobic molecules

- HIGH → LOW concentration gradient

- Facilitated transport

- polar, hydrophilic molecules

- through a protein channel

- HIGH → LOW concentration gradient

- Active transport

- *against* concentration gradient

- LOW → HIGH

- uses a protein pump (requires ATP)



What about large molecules?

- Moving large molecules into & out of cell
 - through vesicles & vacuoles

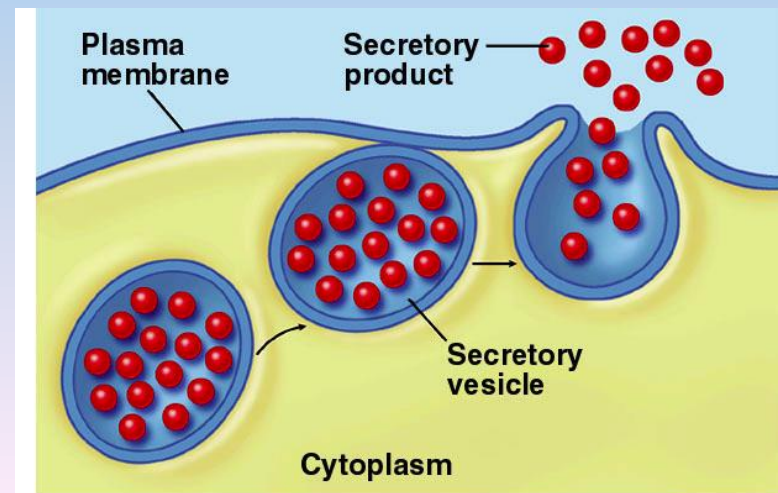
Endocytosis – moving materials INTO the cell

- phagocytosis = “cellular eating”
- pinocytosis = “cellular drinking”
- receptor-mediated endocytosis – Molecules bind to receptors on the outside of the cell membrane.

Exocytosis – moving materials OUT of the cell

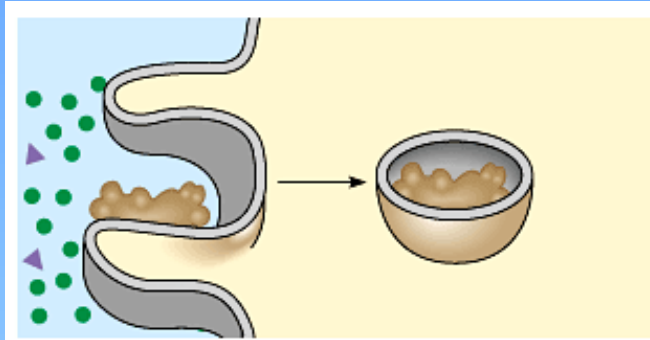
Insulin released from the pancreas into the bloodstream

exocytosis



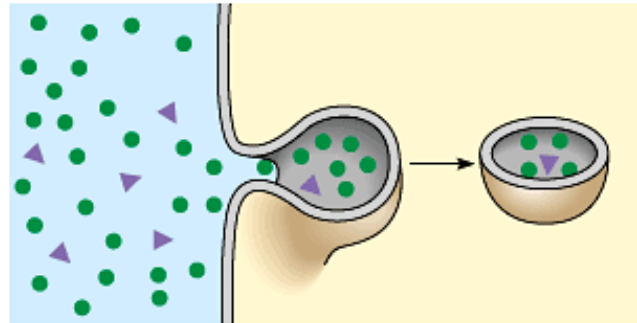
Endocytosis

phagocytosis



White blood cells (macrophages) take in bacteria. fuse with lysosome for digestion.

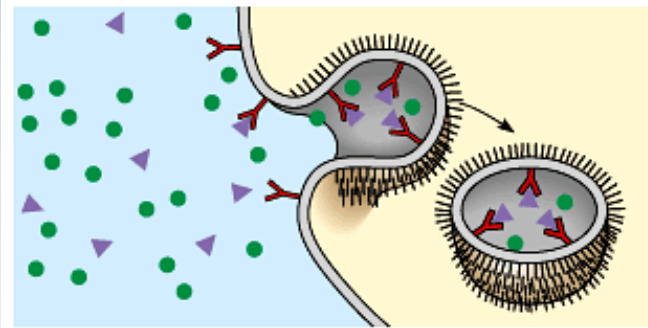
pinocytosis



End products of digestion. non-specific process

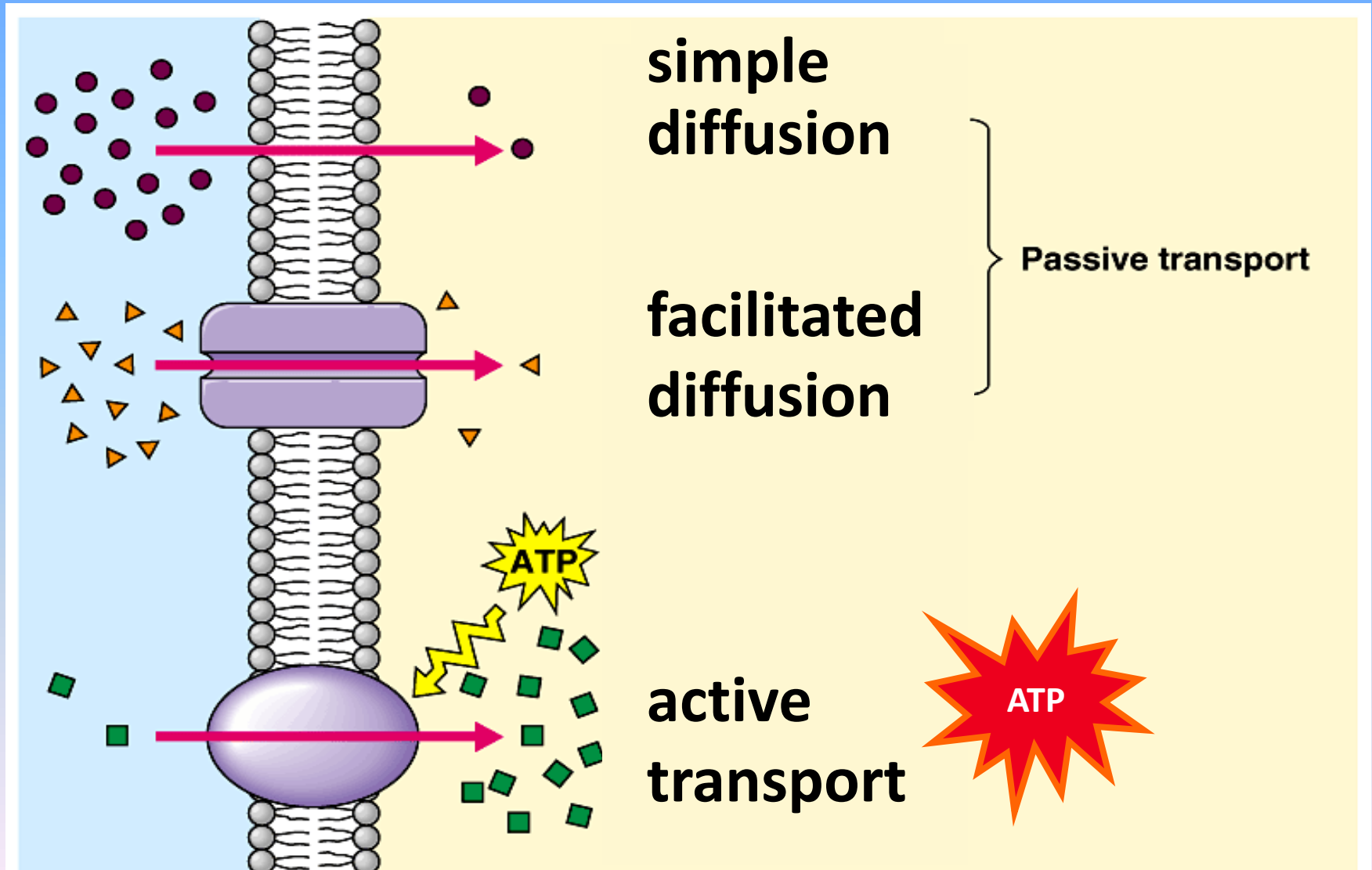
Invagination (pinching) of cell membrane forms a vesicle inside the cell.

receptor-mediated endocytosis

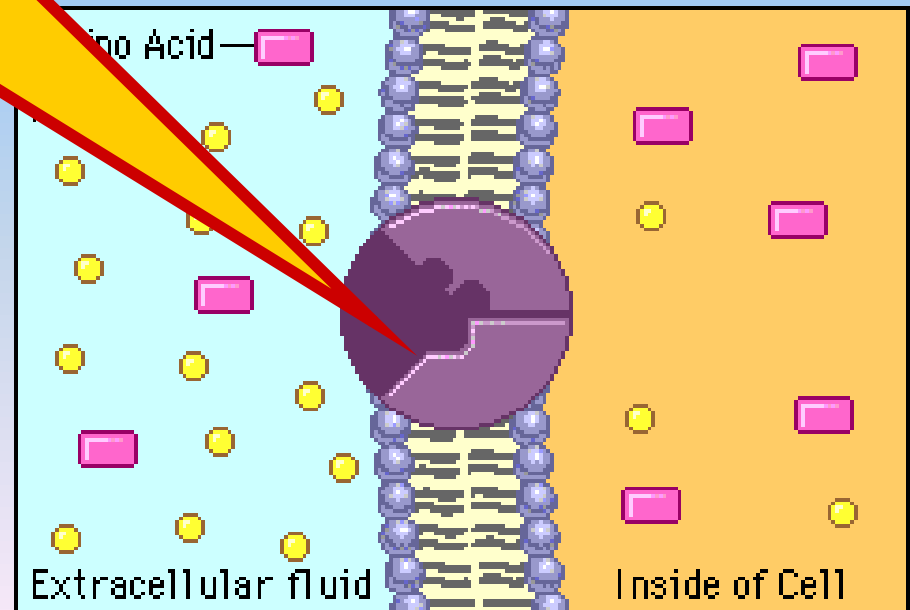


LDL's taken into cell. triggered by molecular signal

Transport summary

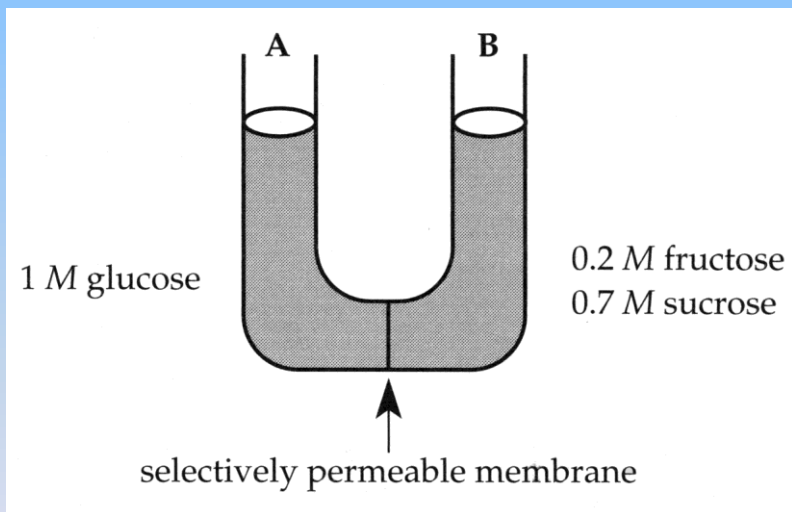


Any Questions??



Review Questions

1. A solution of 1 *M* glucose is separated by a selectively permeable membrane from a solution of 0.2 *M* fructose and 0.7 *M* sucrose. The membrane is not permeable to the sugar molecules. Which of the following statements is correct?



- A. Side A is hypotonic relative to side B.
- B. The net movement of water will be from side B to side A.
- C. The net movement of water will be from side A to side B.
- D. Side B is hypertonic relative to side A.
- E. There will be no net movement of water.

2. You observe plant cells under a microscope that have just been placed in an unknown solution. First the cells plasmolyze; after a few minutes, the plasmolysis reverses and the cells appear normal. What would you conclude about the unknown solute?
- A. It is hypertonic to the plant cells, and its solute can not cross the plant cell membranes.
 - B. It is hypotonic to the plant cells, and its solute can not cross the plant cell membranes.
 - C. It is isotonic to the plant cells, but its solute can cross the plant cell membranes.
 - D. It is hypertonic to the plant cells, but its solute can cross the plant cell membranes.
 - E. It is hypotonic to the plant cells, but its solute can cross the plant cell membranes.